CHEMICAL INVESTIGATION OF THE GENUS RHEEDIA. II. 1 PRENYLATED XANTHONES FROM RHEEDIA GARDNERIANA 2

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ABSTRACT.—Two new prenylated xanthones have been isolated from the root extract of *Rheedia gardneriana*, and their structures have been established as 4',5'-dihydro-1,5,6-trihydroxy-4',5',5'-trimethylfurano (2',3':3,2)-4 (1,1-dimethylprop-2-enyl)-xanthone ($\mathbf{5}$) and 1,5-dihydroxy-6',6'-dimethylpyrano (2',3':6,7)-xanthone ($\mathbf{6}$), respectively.

The genus *Rheedia* (tribe, Clusioideae; family, Guttiferae) has been, so far, scarcely investigated (1), in spite of its close relationship with a widely studied genus—*Garcinia*—which afforded xanthones, polyprenylated benzophenones, and biflavonoids (2). Only recently, three new 1,3,5,6-tetraoxigenated xanthones, namely rheediaxanthone A (1), B (2), and C (3) have been isolated, in addition to macluraxanthone (4), from the root-bark of *Rheedia benthamiana* Pl. and Tr. (3). As a part of a comparative study of the chemistry of a number of brazilian Guttiferae, we examined the roots of *Rheedia gardneriana* Pl. and Tr.

RESULTS AND DISCUSSION

Rheediaxanthone A, B, C, and macluraxanthone (1, 2, 3, and 4, respectively) were again isolated from the benzene extract of the plant and identified by comparison with authentic samples. Moreover, two new prenylated xanthones were obtained and their structures determined. The former, 5, $C_{23}H_{24}O_6$ (MW 396), is a tetraoxigenated xanthone (ir and uv evidence) with an α,α -dimethylallyl chain and an α,α,β -trimethyldihydrofuran ring (1H -nmr evidence), thus isomeric with rheediaxanthone B (2). The identity of the B rings of the two products is substantiated by the uv spectra, with additives, and by the comparable chemical shifts of the H-7 and H-8 protons in xanthones 2-5; it was confirmed by obtaining the two monoacetylderivatives 5a and 5b, in whose 1H -nmr spectra the H-7 and the H-8 protons are shifted downfield alternatively (the data are reported, in comparison with those of the compound 5 and its diacetylderivative 5c, in table 1).

TABLE 1.	Ring B protons resonances (δ , CDCl ₃) in
cc	mpound 5 and its acetylderivatives

									H-7 (d, <i>J</i> =9)	H-8 (d, <i>J</i> =9)
5						_			6.98	7.73
5a									6.97	8.02
5b									7.07	7.73
5c									7.16	8.10

¹Reference (3) can be considered as part I of the series.

²A preliminary communication was presented at the First International Conference on the Chemistry and Biotechnology of Biologically Active Natural Products, Varna, Bulgaria, September 21-26, 1981.

Considering the close accord of the uv spectrum of **5** with that of **1-4** and other 1,3,5,6-tetraoxygenated xanthones, the substitution pattern of the ring A can be only the one indicated in structure **5**. Notably, the propenyl gem-CH₃ resonances in CD_3COCD_3 and C_5D_5N (δ 1.70 and 2.02) are coincident with those of macluraxanthone (**4**, δ 1.75 and 2.01), where the substituent is on C-4 position, and are largely different from those of rheediaxanthone B (**2**, δ 1.57 and 1.76, respectively), where the substituent is on C-2.

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A chemical confirmation of the structure **5** was finally obtained when the xanthone **5** was boiled in HCOOH, yielding the product **5d**, $C_{18}H_{16}O_6$ (MW 328), by loss of the isoprenyl chain. The chemical shift of the generated aromatic proton (H-4) was unchanged in the ¹H-nmr spectrum when the solvent (CD₃COCD₃) was substituted by C_5D_5N (4), as contrasted by the behavior of the isomer **2d**, obtained from rheediaxanthone B (**2**) under the same conditions (3), where the new proton (H-2) is shifted downfield (0.28 ppm). Moreover, only the isomer **2d** showed an immediate bathochromic shift in the uv spectrum upon addition of AlCl₃ (4).

The structure of the latter new xanthone (**6**), $C_{18}H_{14}O_5$ (MW 310), follows from consideration of the spectroscopic data. The uv spectrum relates compound **6** to 1,5,6-trioxygenated xanthones (5). The signals of a *gem*-dimethyl group and of two coupled (J=10 Hz) olefinic protons in the 1H -nmr spectrum and the loss of 15 mass units from the molecular ion in the mass spectrum, to give the base peak, indicate the presence of a 2,2-dimethyl (2H) pyran ring. Further, the shoulder on the longer wave-length band, which is also present in the uv spectra of **1** and **4**, can be considered a typical feature of the chromenoxanthones.

The ¹H-nmr spectrum is completed by the resonances of two hydroxyls, one of which is *peri* to the carbonyl group, and of four aromatic protons. Three of them show up as an AXY system with a coupling pattern very similar (table 2) to that of the ring A protons of buchanaxanthone (1,6-dihydroxy-5-methoxyxanthone) and related compounds (5).

TABLE 2. Ring A protons resonances ^a (δ, CDCl ₃) of 1-hydroxy-xanthones	TABLE 2.	Ring A protons resonances ^a (δ, CDCl ₃) of 1-hydroxy-xanthones
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	H-2	H-3	H-4
Buchanaxanthone	6.67	7.67	7.01
1-Hydroxy-5,6-dimethoxyxanthone	6.80	7.60	7.01
1-Hydroxy-5-methoxyxanthone	6.84	7.65	7.06
1-Hydroxy-5-methoxy-6-propylxanthone		7.56	6.95
Rheediachromenoxanthone	6.80	7.62	6.90

^aH-3 signal is a triplet. H-2 and H-4 show up as *ortho* split doublets with additional *meta* splitting. ^bIn DMSO.

The C_5D_5N influence in the ¹H-nmr spectrum confirms this assignment, with only the H-2 signal being shifted to lower field. Comparison of the resonance of the protons H-8, H-4', and H-5' of **6** and its acetylderivative **6a** with those of **1** and is acetylderivative **1a** established the substitution pattern of the ring B as indicated in the final structure **6**. The compound **6**, for which we propose the name of rheediachromenoxanthone, is the first example of prenylated xanthone with 1,5,6 oxygenation pattern.

EXPERIMENTAL

GENERAL EXPERIMENTAL PROCEDURES.—Elemental analyses were in agreement with molecular formulas. Each mp was determined by means of a Kofler hot plate and is uncorrected. Spectra were recorded with the following instruments: uv, Beckmann Acta III; ¹H-nmr, Varian EM 360; ms, AEI 12; ir, Perkin Elmer 247. Adsorbants used were from E. Merck (plc and tlc) and Macharey-Nagel (cc).

PLANT MATERIAL.—Roots of *Rheedia gardneriana* Pl. and Tr. were collected in northeastern Brazil (Pacatuba, Fortaleza) and identified by José Elias de Paula (Universidade Federal de Brasilia).

EXTRACTION AND SEPARATION OF THE XANTHONES.—Bark (500g) and wood (500g) of the roots gave, by extraction with hot benzene, the same qualitative and quantitative results (7 g each). The crude extract was fractionated on a silica gel column with benzene-ethyl acetate mixtures. Extended cc and plc gave the pure products.

Rheediaxanthone A, (1). 1%, $C_{23}H_{20}O_6$ (MW 392), mp 258-60° (ether-hexane); acetylderivative, (1a): mp 207-9° (CH₂Cl₂-heptane). The identification of rheediaxanthone A (1) and its acetylderivative (1a) was confirmed by comparison with authentic samples from Rheedia benthamiana (3).

Rheediaxanthone B, (2). 15%, $C_{23}H_{24}O_6$ (MW 396), mp 208-10° (benzene-ethyl acetate, $[\alpha]^{22}D=-25$ (c 1.2, acetone); diacetylderivative, (2a): mp 168-9° (CH₂Cl₂-heptane). The identity with rheediaxanthone B (2) and its diacetylderivative (2a) was confirmed by comparison with authentic samples from Rheedia benthamiana. Methylderivatives. Rheediaxanthone B (2) gave, with CH₂N₂, two products, separated on silica gel with benzene: monomethylderivative, (2b): $C_{24}H_{26}O_6$, mp 198-200° (benzene-hexane); δ (CD₃COCD₃): 14.26 (1H, s, ex. D₂O, 1-OH), 7.69 (1H, d, J=9, H-8), 7.10 (1H, d, J=9, H-7), 6.57-6.11 (1H, A part of AXY), 5.07-4.73 (2H, XY part of AXY), 4.50 (1H, q, J=7, H-4'), 4.02 (3H, s, OCH₃), 1.60 (12H, large s, 4 x Me), 1.40 (3H, d, J=7, 4'-Me); λ max (MeOH and AcONa): 259, 286sh, 332; (AlCl₃ after 30' and AlCl₃/HCl): 274,359,420sh. dimethylderivative, (2c): $C_{25}H_{28}O_6$, mp 143-5° (ether-heptane); δ (CCl₄): 13.78 (1H, s, ex. D₂O, 1-OH), 7.75 (1H, d, J=9, H-8), 6.90 (1H, d, J=9, H-7), 6.53-6.08 (1H, A part of AXY), 5.03-4.73 (2H, XY part of AXY), 4.41 (1H, q, J=7, H-4'), 4.01+3.94 (3H+3H, s+s, 5-OCH₃+6-OCH₃), 1.60 (12H, large s, 4 x Me), 1.42 (3H, d, J=7, 4'-Me); λ max (MeOH and AcONa): 244,324; (AlCl₃ after 30' and AlCl₃/HCl): 246,274sh, 350.

Rheediaxanthone C, (3). 1%, $C_{23}H_{24}O_6$ (MW 396), mp 262-4° (ether-heptane), $[\alpha]^{24}D = -22$ (c 0.5, acetone). It was identified as rheediaxanthone C (3) by comparison with an authentic specimen from Rheedia benthamiana (3).

Macluraxanthone, (4). semimicro amount, $C_{23}H_{22}O_6$ (MW 394), mp 180-2° and 204-6° (benzeneethyl acetate). It was identified as macluraxanthone (4) by comparison with an authentic specimen from Rheedia benthamiana (3). $\Delta\delta = \delta$ (C₅D₅N)- δ (CD₃COCD₃): H-8 (+0.35), H-7 (+0.23), H-4' (+0.28), H-5' (-0.08), 6'-Me₂ (-0.07), propenyl-Me₂ (+0.28).

Isorheediaxanthone B, (5). 4',5'-dihydro-1,5,6-trihydroxy-4',5',5'-trimethylfurano (2',3':3,2)-4 (1,1-dimethylprop-2-enyl)-xanthone: 2%, $C_{23}H_{24}O_6$ (MW 396), mp 212-3° (benzene-hexane), $\{\alpha\}^{27}D=+16$ (c 0.6, acetone); δ (CD₃COCD₃): 13.66 (1H, s, ex. D₂O, 1-OH), 7.52 (1H, d, J=9, H-8), 6.92 (1H, d, J=9, H-7), 6.73-6.23 (1H, A part of AXY), 5.36-4.72 (2H, XY part of AXY), 4.43 (1H, q, J=7, H-4'), 1.70 (6H, s, propenyl-Me₂), 1.46+1.23 (3H+3H, s+s, 5'-Me₂), 1.37 (3H, d, J=7, 4'-Me); $\Delta\delta=\delta$ (C₅D₅N)-δ (CD₃COCD₃): H-8 (+0.41), H-7 (+0.30), 5'-Me₂ (+0.04 and +0.07), propenyl-Me₂ (+0.32); ν max (CHCl₃): 3545, 3440, 1657, 1650, 1628, 1589, 1386-1382, 1262, 1188, 114, 1093, 1070, 954, 912, 884; λ max (MeOH): 256 (log € 4.68), 288 (4.25), 332 (4.43); (AcONa): 258, 289, 360; (MeONa): 260, 294, 363; (AcONa/H₃BO₃): 264, 291, 348; (AlCl₃ after 30'): 276, 298, 391; (AlCl₃/HCl): 258, 298, 358; m/z (rel. int.): 396 (M⁺, 44), 395 (3), 381 (100), 379 (9), 369 (3), 367 (1), 365 (2), 355 (3), 351 (3), 339 (5), 327 (1), 325 (8), 323 (2), 313 (7), 311 (3), 299 (2), 297 (2), 295 (1), 285 (2), 281 (1), 269 (1), 198 (1), 183 (3), 69 (4), 41 (5); m*: 366.6 (396→381), 318.2 (396→355), 277.2 (381→325), 257.1 (381→313), 24.4 (69→41).

Acetylderivatives: Isorheediaxanthone B (5) with Ac_2O in pyridine gave three products: the first one was obtained by cc (on silica gel with benzene-ethyl acetate 9:1), whereas the latter two were separated by crystallization and plc.

5,6-Diacetylderivative, (**5c**): $C_{27}H_{28}O_8$, mp 188° (heptane); δ (CDCl₃): 12.94 (1H, s, ex. D₂O, 1-OH), 8.12 (1H, d, J=9, H-8), 7.18 (1H, d, J=9, H-7), 6.48-5.98 (1H, a part of AXY), 5.00-4.64 (2H, XY part of AXY), 4.36 (1H, q, J=7, H-4'), 2.40+2.35 (3H+3H, s+s, 2 x COCH₃), 1.62 (6H, s, propenyl-Me₃), 1.48+1.24 (3H+3H, s+s, 5'-Me₂), 1.35 (3H, d, J=7,4'-Me₂).

5-Monoacetylderivative, (5b): C₂₅H₂₆O₇, mp 203-5° (ether-hexane), green spot with FeCl₃; δ (CDCl₃):

13.20 (1H, s, ex. D_2O , 1-OH), 8.02 (1H, d, J=9, H-8), 6.97 (1H, d, J=9, H-7), 6.58-6.06 (1H, A part of AXY), 5.24-4.70 (2H, XY part of AXY), 4.42 (1H, q, J=7, H-4′), 2.42 (3H, s, COCH₃), 1.65 (6H, s, propenyl-Me₂), 1.43+1.26 (3H+3H, s+s, 5′-Me₂), 1.35 (3H, d, J=7, 4′-Me).

6-Monoacetylderivative. (**5a**): $C_{25}H_{26}O_7$, mp 165° (ether-hexane), violet spot with FeCl₃; δ (CDCl₃): 13.00 (1H, s, ex. D₂O, 1-OH), 7.73 (1H, d, J=9, H-8), 7.07 (1H, d, J=9, H-7), 6.30-5.95 (1H, A part of AXY), 5.45-4.95 (2H, XY part of AXY), 4.23 (1H, q, J=7, H-4'), 2.38 (3H, s, COCH₃), 1.59 (6H, s, propenyl-Me₂), 1.50+1.24 (3H+3H, s+s, 5'-Me₂), 1.30 (3H, d, J=7, 4'-Me).

4′,5′-Dibydro-1,5,6-tribydroxy-4′,5′,5′-trimethylfurano 2′,3′: 3,2)-xanthone. (**5d**): Isorheediaxanthone B (**5**, 30 mg) was refluxed 30 min in HCOOH. After cc purification (silica gel, CH₂Cl₂), 4′,5′-dihydro-1,5,6-trihydroxy-4′,5′,5′-trimethyl furano (2′,3′: 3,2)-xanthone (**5d**, 21 mg) was obtained: $C_{18}H_{16}O_6$ (MW 328), mp 246-8° (ether-hexane), [α]²⁶D=+45 (c 0.2, MeOH); δ (CD₃COCD₃): 12.90 (1H, s, ex. D₂O, 1-OH), 7.50 (1H, d, J=8.5, H-8), 6.87 (1H, d, J=8.5, H-7), 6.22 (1H, s, H-4), 4.43 (1H, q, J=6.5, H-4′), 1.47+1.22 (3H+3H, s+s, 5′-Me₂), 1.37 (3H, d, J=6.5, 4′-Me); $\Delta\delta$ =8 (C_5D_5N)- δ (CD₃COCD₃): H-8 (+0.46), H-7 (+0.35), H-4 (O); λ max (MeOH): 250, 285sh, 328; (AcONa): 258, 280, 356; (AcONa/H₃BO₃): 258, 280sh, 342; (AlCl₃ after 30′): 266, 381; (AlCl₃/HCl): 254, 348; m/z (rel. int.): 328 (M⁺, 25), 313 (M-Me, 100), 298 (9), 295 (7), 285 (16), 269 (6), 257 (6), 149 (33).

Rheediachromenoxanthone, (6). 1,5-dihydroxy-6',6'-dimethylpyrano (2',3': 6,7)-xanthone. 1%, $C_{18}H_{14}O_5$ (MW 310), mp 223-4° (ether-hexane); δ (CD₃COCD₃): 12.90 (1H, s, ex. D₂, 1-OH), 8.40 (1H, s, ex. D₂O, 5-OH), 7.62 (1H, t, J=8.5, H-3), 7.43 (1H, s, H-8), 6.94 (1H, dd, J=8.5 and 1.5, H-4), 6.70 (1H, dd, J=8.5 and 1.5, H-2), 6.55 (1H, d, J=10, H-4'), 5.87 (1H, d, J=10, H-5'), 1.50 (6H, s, 6'-Me₂); $\Delta \delta = \delta$ (C₅D₅N)−δ (CD₃COCD₃): H-8 (+0.30), H-3 (-0.09), H-4 (-0.04), H-2 (+0.22), H-4' (-0.05), H-5' (-0.12), 6'-Me₂ (-0.17); ν (CHCl₃): 3540, 1642, 1605sh, 1600sh, 1588, 1393, 1384, 1133, 1072, 884, 814; λ max (MeOH and AcONa): 262 (log € 4.57), 282 (4.59), 312sh, (4.08), 338sh (4.04), 372sh (3.96); (AlCl₃ and AlCl₃/HCl): 272, 294, 370, 428; m/z (rel. int.): 310 (M⁺, 29), 309 (10), 295 (M-Me, 100), 266 (1), 181 (1), 165 (M/2, 2), 147.5 (M - Me/2, 11), 133.5 (M - Me - CO/2, 2.5); m*: 280.7 (310→295).

Diacetylderivative, (6a): Rheediachromenoxanthone (6) with Ac_2O in pyridine overnight gave a diacetylderivative (6a), purified on silica gel (benzene- CH_2Cl_2 mixtures): $C_{22}H_{18}O_7$, mp 183-5°, H-2 (ether-hexane); δ (CDCl₃): 7.69, H-4 (1H, s, H-8), 7.52 (1H, t, J=8.5, H-3), 7.28 (1H, dd, J=8.5 and 1.5), 6.92 (1H, dd, J=8.5 and 1.5), 6.36 (1H, d, J=10, H-4'), 5.70 (1H, d, J=10, H-5'), 2.47 (6H, s, 2 x COCH₃), 1.47 (6H, s, 6'-Me₂).

Methylderivatives: Rheediachromenoxanthon (**6**) with CH₂N₂ afforded a monomethyl-(**6b**) and a dimethylderivative-(**6c**), separated on silica gel with hexane-ethyl acetate mixtures; monomethylderivative: C₁₉H₁₆O₅, mp 122-3° (ether); δ (CDCl₃): 12.70 (1H, s, ex. D₂O, 1-OH), 7.79 (1H, s, H-8), 7.73 (1H, t, J=8.5, H-3), 6.90 (1H, dd, J=8.5 and 2, H-4), 6.70 (1H, dd, J=8.5 and 2, H-2), 6.35 (1H, d, J=10, H-4'), 5.68 (1H, d, J=10, H-5'), 3.97 (3H, s, OCH₃), 1.48 (6H, s, 6'-Me₂): dimethylderivative: C₂₀H₁₈O₅, mp 123-4° (ether-hexane); δ (CDCl₃): 7.59 (1H, s, H-8), 7.49 (1H, t, J=8.5, H-3), 7.02 (1H, dd, J=8.5 and 2, H-4), 6.72 (1H, dd, J=8.5 and 2, H-2), 6.37 (1H, d, J=10, H-4'), 5.63 (1H, d, J=10, H-5'), 3.97 (6H, s, 2 x OCH₃), 1.52 (6H, s, 6'-Me₂).

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